

Oxygen Safety

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ACADEMIC-INDUSTRY 2023 LIQUID ROCKET SYMPOSIUM

Introduction

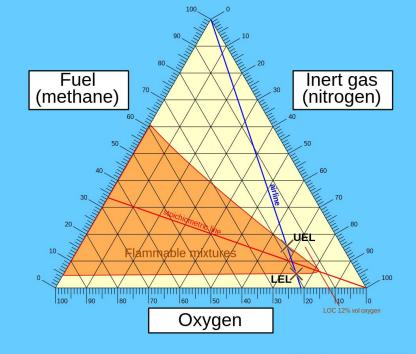
- Important References
- Oxygen and Combustion
- Material Compatibility
- Oxygen Rich Environments
- Ventilation
- Oxygen Cleanliness

Important References

- ASTM G88
- ASTM MNL 36
- NASA 1740.15

Oxygen and Combustion

- Once a quantity of flammable fuel-oxygen mixture reaches autoignition temperature, it will ignite
 - Volatile fuels can be lit by a small initial heat source in a runaway reaction
- Oxygen fuels combustion, and pure O2 is more capable of supporting combustion



HEALTH HAZARD FIRE HAZARD Flash Points 4 - Deadly 4 - Below 73° F 3 - Extreme Danger 3 - Below 100° F 2 - Hazardous 2 - Below 200° F 1 - Slightly Hazardous 1 - Above 200° F 0 - Normal Material 0 - Will Not Burn Acid.....ACID 4 - May Detonate 3 - Shock and Heat Alkali......ALK **May Detonate** Corrosive.....COR 2 - Violent Chemical Oxidizer.....OX Radiation Hazard... 1 - Unstable if Heated 0 - Stable Use No Water.... ₩ SPECIFIC HAZARD **INSTABILITY HAZARD**

NFPA Rating Label by SafetySign.com

Material Compatibility



- Aluminum and Steel can sometimes ignite
 - Small particles or freshly eroded surfaces
 - Titanium is highly combustible with GOX and LOX
- Stainless steels, copper and nickel alloys, and PTFE are generally acceptable, but not always.

- Wood and plastics are also fuels
- Lubricants, Oils, and Grease are combustible, and may be found inside valves and components
 - Use oxygen compatible lubricants
- Kerosene fuels contaminate surfaces
- Natural fibers, though combustible, will not melt to your skin like synthetics

Oxygen and Combustion

- Ignition sources are not limited to a spark, consider:
 - Temperature: Hot gas or materials can transfer thermal energy to a combustible mixture
 - Impact and friction: Both generate heat
 - Static Electricity: Generates sparks that produce heat
 - Compression Heating: Compressing gasses generates heat



- : $Heat = c_p m \Delta temp$
- $\frac{P_1V}{T_1} = \frac{\uparrow P_2V}{\uparrow T_2}$ (ideal gas)

Oxygen Rich Environments

- LOX creates an oxygen rich environment where it ends up
 - Not just in the motor, but also in fabrics, on surfaces, in the air
 - LOX leaking through valve stems and seals generates GOX
 - Humans and other organic materials



- Concentrated oxygen supports combustion easier
 - Case Study: Apollo 1
- Oxygen and fuel rich volumes can detonate
 - Case Study: Boomie Zoomie B

Ventilation



- Know where to place your ullage vents.
 Get the gasses away from the rocket and from each other
 - Vent oxygen and fuel and different heights. The wind may carry them horizontally
 - Use standoff tubes to vent gasses outside of the airframe

- Don't trap fuel and oxygen vapors inside partially enclosed volumes
- LOX may leak from valve stems or fittings and generate oxygen gas
 - Will also occur with fuels such as LNG

Oxygen Cleanliness

- Systems can be kept fairly safe for oxygen use if you keep your systems clean:
 - Keep equipment and hardware off dirt
 - Clean all valves, fittings, and tubing
 - Clean fuel or lubricant spills promptly before you forget or track it elsewhere
 - Deburr parts and maintain passivation



- Purge with inert gas to help remove contaminants from oxygen systems and oxygen from fuel systems
- Cleaning procedures will be covered in a later demonstration
- ALWAYS CAP AND PLUG OR TAPE